Alkyl(phenylthio)ketenes as Synthetic Equivalents of Monoalkylketenes: A Concise General Route To 3-Alkyl 8-Lactams as Carbapenem Building-Blocks.

Claudio Palomo*, Fernando P. Cossio, José M. Odriozola, Mikel Olarbide, and Jesús M. Ontoria. Departamento de Química Aplicada, Unidad de Química Orgánica, Facultad de Ciencias Químicas.Universidad del Pais Vasco Apdo. 1072, 20080 San Sebastián. Spain.

Summary: The dehydrochlorination of α-phenylthicalkanoyl chlorides with triethylamine in the presence of imines produced a high yield formation of α -phenylthio β -lactams which upon desulfuration furnished a wide variety of 3-

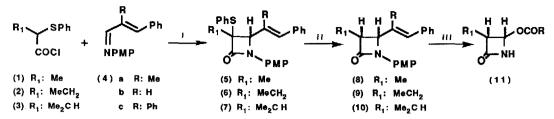
Appropriately substituted monocyclic 3-alkyl B-lactams are important starting materials for the preparation of a wide variety of bicyclic β-lactam antibiotics (ie, carbapenems, olivanic acids, penems and exapenems)¹. Recently we have reported on the utility of the α -bromoester-imine condensation to the synthesis of appropriately substituted 3-alkyl B-lactams². Among many other methods for the synthesis of monocyclic B-lactams³, the ketene-imine cycloaddition reaction⁴, or its equivalent, the acid chlorideimine method, has proven to be an exceedingly effective route for the construction of the azetidinone ring⁵. However, the direct preparation of 3-alkyl B-lactams from monoalkylketenes, generated from their corresponding acid chlorides, is often limited in scope⁶. Although some exceptions have recently been appeared⁷, no general method to the synthesis of 3-alkyl β -lactams from the ketene-imine approach has been described.

We report here a concise general route to 3-alkyl B-lactams as carbapenem building blocks which involves the use of alkyl(phenylthio)ketenes as synthetic equivalents of monoalkylketenes (Scheme I). Preparation of the 3-methyl B-lactam 8a was examined first. Thus α -(phenylthio)propanoyl chloride 1 with the imine 4a, derived from α -methylcinnamaldehyde and p-anisidine⁸, in methylene chloride as solvent afforded, under standard conditions⁹, the expected 3-methyl-3-phenylthio β -lactam 5 in 97% yield as a mixture of cis and trans isomers in a ratio 80:20 respectively¹⁰. When this mixture of isomers was then treated with a slightly excess of tributyltin hydride under azoisobutyronitrile (AIBN) catalysis¹¹

Compound ^b	Solvent	Yield, % ^C	cis:trans ^d	Compound ^b	Solvent	Yield, % ^C	cis:trans ^d
<u>5a</u>	CH2CI2	97	80:20	<u>5a</u>	Benzene	95	91:9
<u>5a</u>	CH3CN	80	72:28	<u>5c</u>	CH ₂ Cl ₂	84	100:0
<u>6a</u>	CH ₂ Cl ₂	86	84:16	<u>6a</u>	Benzene	90	91:9
<u>6a</u>	CH3CN	85	77:23	<u>6b</u>	CH2CI2	24	64:36
<u>6b</u>	Benzene ^e	72	77:23	<u>6b</u>	CH ₃ CN ^e	75	57:23
<u>6c</u>	CH2Cl2	90	100:0	<u>6c</u>	Benzene	92	100:0
<u>7a</u>	CH ₂ Cl ₂	95	62:38	<u>7a</u>	Benzene	97	86:14
<u>7a</u>	CH3CN	90	57:43				

Table I. Solvent effect on the stereochemical outcome in the cycloaddition step between imines and α phenylthioalkanovl chloridesa.

^a Reactions conducted on 10 mmol scale, by addition of the acid chloride to a solution of the imine and triethylamine. b All compounds are racemic and were characterized by their physical properties and analytical data. ^c Yields based on weight of isolated product by column chromatography. ^d Determined by 300 MHz nmr spectroscopy: the H-4 proton in the cis isomer appears at lower field than that in the trans isomer. e Addition of the acid chloride at 65°C, then r.t.



SCHEME I. Reagents and Conditions: i, NEt₃, CH₂Cl₂, r.t. 20-24h.; ii, n-Bu₃SnH, AIBN, toluene, 1-2h.; iii, ref. 2. PMP: p-methoxyphenyl group.

in refluxing toluene for 1.5h, a 92:8 ratio of cis and trans isomers of 8a was efficiently produced in 80% yield. Crystallization from cyclohexane afforded pure cis-8a [mp. 113-115°C; 8 3.60 (dq, H-3), 4.60 (d, J=5.4Hz, H-4)]. The extension of this method to the preparation of the corresponding 3-ethyl and 3isopropyl analogues is shown in Table I. The following results are particularly noteworthy: first, the stereoselectivity of the cycloaddition step could be notably improved by changing the starting imines 4a and 4b to the more bulky Schiff base 4c12; second, the isomer distribution of these 3-alkyl 3-phenylthio Blactams was found to be strongly dependent upon the solvent media, and the best stereochemical outcome in the cycloaddition step could be obtained when benzene was the solvent of choice. These 3-alkyl-3phenylthio B-lactams, upon treatment with tributyltin hydride reagent under AIBN catalysis provided the corresponding 3-alkyl B-lactams in excellent yields¹³. As can be seen in Table II, a mixture of cis and trans isomers was generally produced, in which the cis isomer predominate. This high degree of cis stereoselectivity could be explained by assuming that the hydride attack takes place preferentially at the less hindered face of the starting α -phenylthio β -lactam and in the case of <u>6c</u>, that incorporates a more bulky C4-substituent, only the corresponding cis isomer of 9c was formed. All of these 3-alkyl β-lactams could be further elaborated by established protocols to furnish important carbapenem precursors, like 111.

Compound ^b	Yield, % ^C	cis:trans ^d	Compoundb	Yield, % ^C	cis:trans ^d	
<u>8a</u>	80	92:8	<u>9a</u>	97	92:8	
<u>9b</u>	87	76:24	<u>9c</u>	97	100:0	
<u>10a</u>	91	88:12				

Table II. 3-alkyl ß-lactams prepareda.

^a The reduction was performed on a mixture of cis and trans α -alkyl(thiophenyl) β -lactams prepared in methylene chloride as solvent. Reactions conducted on 10 mmol scale in refluxing toluene by using n-Bu₃SnH (1.2 equiv.) and AIBN as catalyst (0.1 equiv.). ^b All compounds are racemic mixtures and were characterized by their physical properties and analytical data. ^c Yields based on weight of isolated product by column chromatography. ^d Determined by 300 MHz nmr spectroscopy: J_{3.4} \approx 5 Hz (cis isomer).

In view of the above results, we tested the scope of our method by examining the reaction between α -phenylthiobutanoyl chloride 2 and Schiff bases 12, that incorporate other suitable substituents for further chemical elaboration. Results of this study are summarized in Table III to illustrate the efficiency, the scope and the applicability of the present method. As can be seen from the results in Table III, imines bearing alkyl substituents at the imine nitrogen atom or carbonyl groups at the imine carbon atom, which can not be used under standard enolate-imine reaction conditions¹⁴, provided the corresponding 3-alkyl-

imine	∋ <u>12</u> b	produc	ct <u>13</u> 0		imine	<u>12</u> b	produ	uct <u>13</u> 0
R ₁	R ₂	Yield, % ^d	cis:trans ^f		R ₁	R ₂	Yield, % ^d	cis:trans ^f
a) PhCO	PMP	95	19:819	(b)	PhC=C	PMP	90	60:40
c) PhCH=CM	e CH ₂ CO ₂ Me	74 ⁰	92:8	(d)	CO ₂ Me	PMP	89	7:93

Table III. (±)PS-5 carbapenem building blocks prepared^a.

^a Reactions conducted on 10 mmol scale, by addition of the acid chloride to a solution of the imine and triethylamine in benzene as solvent unless otherwise stated. ^b Prepared by standard procedures. ^c All compounds are racemic and were characterized by their physical and analytical data. ^d Yields based on weight of isolated product by column chromatography. ^e In this case the imine was prepared "in situ" and not isolated. ^f Determined by 300 MHz nmr spectroscopy. ^g Reaction performed in methylene chloride as solvent.

3-phenylthio β -lactams <u>13</u> in high yields, as valuable intermediates for the synthesis of the (±)PS-5 carbapenem antibiotic¹. For instance, when the β -lactam <u>13c</u> was allowed to react with tributyltin hydride under AIBN catalysis, the β -lactam <u>14c</u> was produced in 80% yield as a mixture of cis and trans isomers in a ratio 82:18 respectively. In a similar way, the β -lactam <u>13d</u> furnished a 71:29 ratio of cis and trans isomers of <u>14d</u>. Attempted preparation of the β -lactam <u>14c</u> directly from butanoyl chloride and the imine <u>12c</u> was unfruitful¹⁵; the α -bromoester-imine condensation gave neither the expected β -lactam <u>14c</u> nor <u>14d</u>. The use of α -haloalkylketenes¹⁶ as synthetic equivalents of monoalkylketenes was found to be less efficient (Figure I). Reaction between α -halopropanoyl chloride in the presence of triethylamine afforded the β -lactam (15) in 40% yield together with the pyridone (17) in 52% yield. When the reaction was examined with 2-bromopropanoyl chloride, the yield of the expected β -lactam (16) decreased up to 25% and the dehydrobrominated [4+2] cycloadduct (17) was obtained as main product in 70% yield¹⁷.

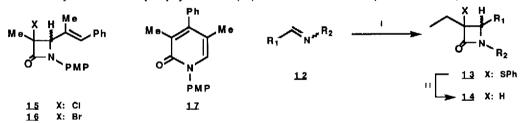


Figure I. Products obtained in the reaction SCHEME II. Reagents and Conditions: i, MeCH₂CH(SPh)between α -halopropanoyl chlorides and the imine COCI, solvent, r.t. 20-24 h.; ii, n-Bu₃SnH, AIBN, toluene, (4a). <u>17</u> m.p. 106-107.5°C (cyclohexane). reflux.

From the results reported here it is clear that a wide variety of carbapenem building blocks can be obtained by means of alkyl(phenylthio)ketenes as synthetic equivalents of monoalkylketenes¹⁸. Further studies on the application of this methodology to the synthesis of optically active carbapenem compounds are now underway in our laboratory.

Acknowledgement: The present work has been supported by Comisión Interministerial de Ciencia y Tecnología (Project FAR: 88-0393). A grant from the Ministerio de Educación y Ciencia and Eusko Jaurlaritza to J.M. Odriozola, M. Oiarbide and J.M. Ontoria are gratefully acknowledged.

References

1.- For recents reviews, see: (a) T. Nagahara, T. Kametani, Heterocycles 1987, 25, 729. (b) G.I. Georg In

"Studies in Natural Product Chemistry", Rahman, A-ur, Ed.; Elsevier Science: Amsterdam 1989, in press.

- J.M. Odriozola, F.P. Cossío, C. Palomo, J. Chem. Soc.; Chem. Commun. 1988, 809. F.P. Cossío, J.M. Odriozola, M. Oiarbide, C. Palomo, J. Chem. Soc.; Chem. Commun. 1989, 74. C. Palomo, F.P. Cossío, A. Arrieta, J.M. Odriozola, M. Oiarbide, J.M. Ontoria, J. Org. Chem., in press.
- 3.- For reviews, see: (a) A.K. Mukerjee, R.C. Srivastava, Synthesis 1973, 373. (b) N.S. Isaacs, Chem. Soc. Rev. 1976, <u>76</u>, 181. (c) A.K. Mukerjee, A.K. Singh, Tetrahedron 1978, <u>34</u>, 1731. (d) G.A. Koppel in "Small Ring Heterocycles, Azetidines, β-Lactams, Diazetidines and Diaziridines", Hassner, A. Ed.; Wiley: New York, 1983, Chapter 2.
- 4.- a) H. Staudinger, Llebigs Ann. Chem. 1907, <u>356</u>. 51. For reviews, see: b) W.E. Hanford, J.C. Sauer, Org. React. 1946, <u>3</u>, 108. c) S. Patai, Ed. "The Chemistry of Ketenes and Allenes and Related Compounds", Parts 1 and 2, Wiley: New York, 1980.
- For recent reviews, see: (a) T.R. Govindachari, P. Chinnasamy, S. Rajeswari, S. Chandrasekaran, M.S. Premila, S. Natajaran, B.R. Pai, Heterocycles 1984, <u>22</u>, 585. (b) J.S. Sandhu, B. Sain, Heterocycles 1987, <u>26</u>, 777.
- 6.- This fact is probably due to the inherent unstability associated with aldoketenes, see for example, J.C. Sauer, J. Am. Chem. Soc. 1947, <u>69</u>, 2444 and C.C. McCarney, R.S. Ward, J. Chem. Soc.; Perkin Trans I 1975, 1600 and R.N. Lacey in " The Chemistry of Alkenes", S. Patai, Ed., Interscience: New York, 1964. See, also R.S. Ward in ref. 4c, part 1 p. 231.
- B. Alcaide, G. Dominguez, G. Escobar, V. Parreno, J. Plumet, Heterocycles 1986, <u>24</u>, 1579. B. Ernest, D. Bellus, DE. 3620467AI, 1987, Chem. Abstr. 1987, <u>106</u>, 176045q. D.M. Tschaen, L.M. Fuentes, J.L. Lynch, W.L. Laswell, R.P. Volante, I. Shinkai, Tetrahedron Lett. 1988, <u>29</u>, 2779.
- 8.- A. Arrieta, B. Lecea, F.P. Cossío, C. Palomo, J. Org. Chem. 1988, 53, 3784.
- 9.- M. Ishida, T. Minami, T. Agawa, J. Org. Chem. 1979, 44, 2067.
- 10.- The assignment of the relative stereochemistry was performed on the basis of 300MHz ¹H-NMR spectroscopy by measurement of the nuclear Overhauser enhancement on the signal corresponding to the C₄-H methine protons previous saturation of the methyl group signal on each isomer.
- C.G. Gutierrez, L.R. Summerhays, J. Org. Chem. 1984, <u>49</u>, 5206. For recent reviews on tributyltin hydride, see: M. Pereyre, J.P. Quintard, A. Rahn in " Tin in Organic Synthesis", Butterworths: London, 1987 and W.P. Newman, Synthesis 1987, 665.
- 12.- J. M. Aizpurua, F.P. Cossio, B. Lecea, C. Palomo, Tetrahedron Lett. 1986, 27, 4359.
- 13.- Representative data: cis-(9a) m.p. 99-100°C (EtOH); δ 3.39 (ddd, J=8.7Hz, J'=6Hz, H-3), 4.59 (d, J=6Hz, H-4). cis-(9b); δ 3.34 (td, J=7.8Hz, J'-6Hz, H-3), 4.67 (dd, J=7.9Hz, J'=6Hz, H-4). trans-(9b); δ 3.02 (td, J=7.3Hz, J'=1.8Hz, H-3), 4.27 (dd, J=8.4Hz, J'=1.8Hz, H-4). cis-(9c); δ 3.05 (td, J=6.6Hz, J'=2Hz, H-3), 4.02 (d, J=6.6Hz, H-4). cis-(10a) m.p. 76-78°C (hexane); δ 2.91 (dd, J=8.1Hz, J'=2.4Hz, H-3), 4.30 (d, J=2.4Hz, H-4).
- 14.- G.I. Georg, J. Kant, H.S. Gill, J. Am. Chem. Soc. 1987, 109, 1129.
- 15.- A.K. Bose, Y.H. Chiang, M.S. Manhas, Tetrahedron Lett. 1972, 4091. Aliphatic acids in the presence of activating agents and triethylamine neither react well with imines to produce β-lactams, see: A. Arrieta, B. Lecea, C. Palomo J. Chem. Soc.; Perkin Trans I 1987, 845.
- For a review on halogenated ketenes, see: W.I. Brady, Tetrahedron Lett. 1981, <u>37</u>, 2949. For an application in β-lactam chemistry, see: P. Lombardi, M. Colombo, A. Crugnola, G. Franceschi, GB 2 144 419. A, 1985; Chem Abstr. 1985, <u>103</u>, 53864m.
- For a discussion on the factors that control the stereoselectivity and periselectivity of the cycloaddition between ketenes and imines, see: H.W. Moore, G. Hughes, K. Srinivasachar, M. Fernandez, N.V. Nguyen, D. Schoon, A. Tranne J. Org. Chem. 1985, <u>50</u>, 4231.
- 18.- For a recent review on the applications of ketenes, see: B.B. Snider, Chem. Rev. 1988, 88, 793.

(Received in UK 10 July 1989)